



Written on 26 July 2023



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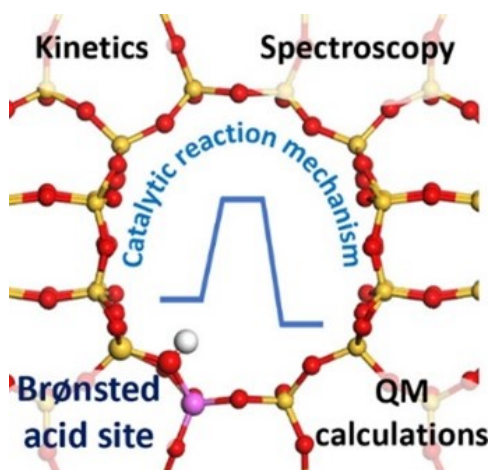


News

Fundamental Research

Catalysis and reaction kinetics

Researchers from IFPEN's Catalysis, Biocatalysis and Separation division have recently published a state-of-the-art report on transformations catalyzed by Brønsted acids in zeolites. Published in the journal Chemical Reviews, this body of knowledge illustrates IFPEN's growing involvement in the energy and ecological transition.



State-of-the-art article published

Published in the special issue [Bridging the Gaps: Learning from Catalysis across Boundaries](#) of the journal Chemical Reviews, the article presents an overview of the current state of knowledge on **the mechanisms of reactions catalyzed by Brønsted acids in zeolites**. This subject has been the focus of major research efforts by the scientific community for decades, and has many potential applications.

As well as being of practical interest to those working in the field of chemical and process innovation, this publication illustrates IFPEN's involvement in research challenges that are vital to transforming the industry in response to climate change.

It also reflects the recognition of IFPEN teams by the international scientific community

Catalysis: a field of chemistry at the heart of the ecological and energy transition

At the heart of IFPEN's R&I, catalysis is regularly in the spotlight, with [researchers receiving awards for their work](#). Catalysis offers a wide range of benefits for a more sustainable energy future. For example, **it lowers the activation energies required to initiate chemical reactions**, and **improves their speed and selectivity** (reducing the formation of other molecules/waste). It also facilitates **the biobased production of fuels and chemical intermediates**. In addition, it is involved in **processes that convert CO₂ into fuel**.

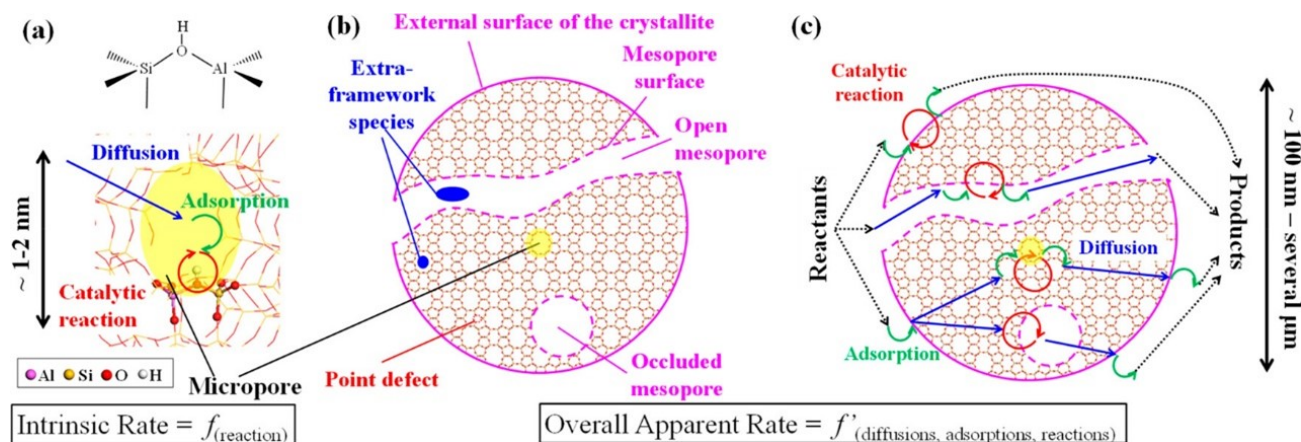
An innovative molecular approach

The authors have adopted an innovative molecular approach, bringing together both experimental and computational methods. The elements reported thus emanate from several types of techniques: **detailed kinetic studies, in situ and operando spectroscopies, quantum chemical calculations**. These include the identification of relevant intermediates and transition states, and the respective kinetics of their formation and transformation into products.

Detailed characterization of sites and phenomena at work

When it comes to zeolite catalysis, **the structure, location and accessibility of active sites** is a key consideration. This aspect has been extensively addressed, providing insights into the nature and acidity of possible active sites, as well as the role of the various phenomena at work: **adsorption, diffusion, reaction kinetics**.

The figure below outlines the most relevant concepts in this field, which are important for understanding the approaches undertaken, the knowledge obtained to date and the remaining challenges for understanding the mechanisms catalyzed by zeolites in their protonic form.



(a) Model of Brønsted acidic zeolite, restricted to the micropore, with consideration of diffusion, adsorption, and reactions steps. (b) Scheme of defect and finite size nature of an intricate zeolite crystallite. A spherical crystallite model is arbitrarily chosen. (c) Integration of the variety of sites and of steps for the definition of the overall apparent reaction rate. Copyright 2023 American Chemical Society

Important reactions for industrial applications

A large part of the paper is devoted to the description of atomic-scale mechanisms for reactions of great industrial importance, such as:

- conversion of alkenes, alkanes and aromatic molecules;
- dehydration reactions of alcohols and polyhydroxylated molecules;
- carbonylation reactions;
- reactions involved in the conversion of methanol to olefins.

Reference :

(1) Chizallet C., Bouchy C., Larmier K., Pirngruber G. (2023). Molecular Views on Mechanisms of Brønsted Acid-Catalyzed Reactions in Zeolites, Chem. Rev. 2023, 123, 9, 6107–6196 30, 2023, <https://doi.org/10.1021/acs.chemrev.2c00896>

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